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c. only carbon. d. only carbon and hydrogen. 2. The number of covalent bonds a carbon atom can form with other atoms is 3. A covalent bond formed when two atoms share two pairs of electrons is called a a. -single bond. b. double bond. 4. The breakdown of a polymer involves a. hydrolysis.

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Section 2-3 Carbon Compounds. (pages 44-48) This section explains how the element carbon is able to form millions of carbon, or organic, compounds. It also describes the four groups of organic compounds found in

living things. The Chemistry of Carbon.

Macromolecules The Chemistry of Carbon

Section 3-2 VOCABULARY REVIEW 1. An organic compound is a compound containing carbon atoms covalently bonded to other carbon atoms and to other elements. Examples: any carbon-containing compound, such as benzene, ethanol, glycerol, glucose, fructose, sucrose, ATP, and ADP. 2. A functional group is a cluster of atoms in a com-

SECTION 3-3 REVIEW MOLECULES OF LIFE

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Section 2-3 2-3 Carbon Compounds

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